3.18 The free energy as an example for a thermodynamic potential

The **free energy** is the corresponding thermodynamic potential for the thermal contact of section 3.15. Therefore the corresponding coordinates are V, N, and T. It is well known that

$$F(V, N, T) = U(V, N, T) - S(V, N, T)T$$
(3.32)

 $\quad \text{and} \quad$

$$dF = \mu dN - p dV - S dT \tag{3.33}$$

Mathematical interpretation: total differential, partial derivative

$$\frac{\partial F}{\partial N}\Big|_{V,T} = \mu \quad \left| \begin{array}{c} \frac{\partial F}{\partial V} \Big|_{N,T} = -p \\ \left| \begin{array}{c} \frac{\partial F}{\partial T} \Big|_{V,N} = -S \end{array} \right|$$

Physical interpretation: gradient, forces

μ : "force" changing the	-p: "force" changing the	-S: "force" changing the
particle number	volume	temperature

here: V, N, T: generalized coordinates $-p, \mu, -S$: generalized forces

For "normal" physical quantities:

Coordinates: extensive parameters

Forces: intensive parameters

Forces and coordinates can be exchanged. There exists no fundamental difference between these parameters since they are equivalent and just depend on the choice of the thermodynamic contact.