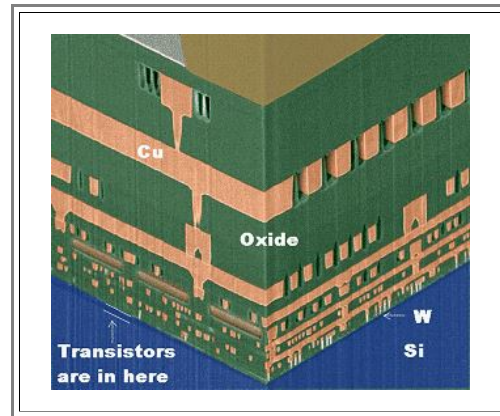


3.6 Summary

3.6.1 Summary to: 3. Thin Films

Semiconductor technology is almost synonymous with thin film technology.

- A thin film is always adhering to a substrate and (at least originally) continuous.
- Thin films may still be found in the product or may have been "sacrificed" during the making of the product.
- An **IC** is a study of thin films in and on the **Si** substrate.
- The same is true for pretty much every semiconductor product.



Thin always means "thin" relative to some intrinsic (internal) length scale. Examples are:

- Structural length scales.
- Wavelength and Interaction length scales.
- Transport parameter length scales.
- Electrical scales.

- | | |
|---|--|
| <ul style="list-style-type: none"> • Dimensions $d_{x, y, z}$ • Grain size d_{grain} • Lattice constants a_0 • λ radiation (light, IR, UV) • Absorption depths • Mean free path lengths. • Diffusion length | <ul style="list-style-type: none"> • SCR width d_{SCR} • Debye length d_{Debye} • Critical thickness d_{crit} for electrical break down • Critical thickness d_{tu} for tunneling |
|---|--|

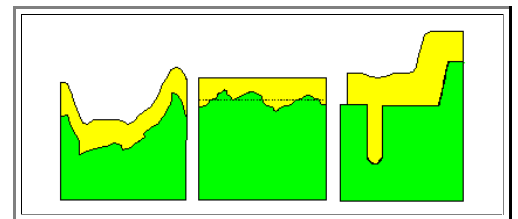
There are many thin film applications outside of semiconductor technology:

- Optical, electrical, chemical, mechanical, magnetic technologies use thin films.

Thin films have other spatial properties besides their thickness.

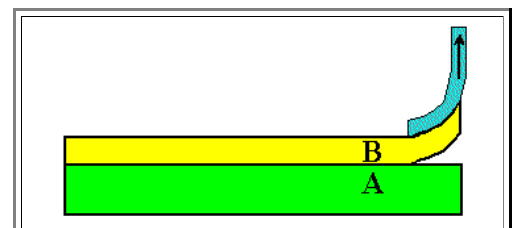
- Interface roughness and surface roughness R defined by their "root mean square":

$$R = \frac{1}{N} \sum_{i=1}^N |z_i|$$



Thin films adhere to their substrate.

- A direct measure of adhesion is the interfacial energy γ_{AB} between film **A** and substrate **B**.
- The phase diagram provides some guideline. Complete miscibility=good adhesion, (eutectic) decomposition=(?) low adhesion. Calculations of γ are difficult.
- Full adhesion can only be obtained for films grown on a substrate. Adhesion energies can be measured.



Generally, there will be stress σ and strain ϵ in a thin film and its substrate.

- A major source of strain is the difference of the thermal expansion coefficients α .

Stress and strain in thin films can be large and problematic!

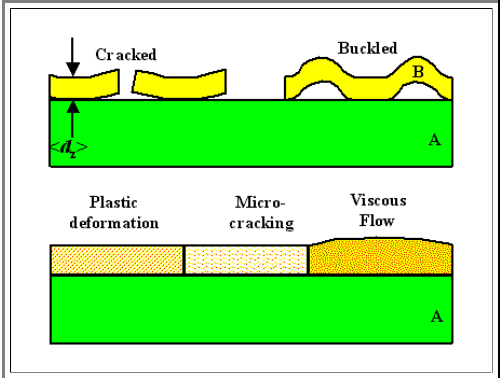
$$\epsilon_{TF} = \Delta T \cdot \Delta \alpha$$

$$\sigma_{TF} = Y \cdot \Delta T \cdot \Delta \alpha$$

Stress in thin film may relax by many mechanisms, and this might be good or bad:

- Cracking or buckling.
- plastic deformation.
- Viscous flow.
- Diffusion.
- Bending of the whole system (Warpage).

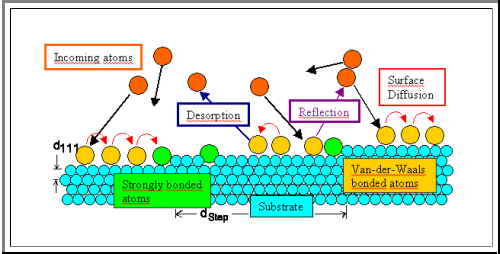
Warpage can be a serious problem in semiconductor technology.



Deposition of a thin layer must start with a "clean" substrate surface on which the first atomic / molecular layer of the film must nucleate.

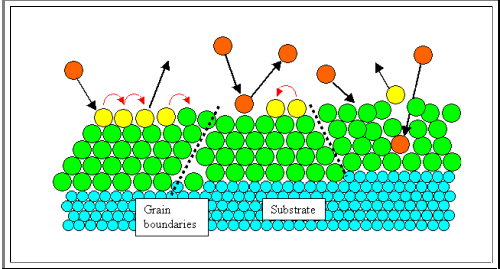
There are many possible interactions between the substrate and "first" incoming atoms.

- As the interaction energy goes up we move from "some" absorption to physisorption (secondary bonds are formed) to chemisorption (full bonding)
- The sticking coefficient is a measure of the likelihood to find an incoming atom in the thin film forming.
- Immobilization by some bonding is more likely at defects (=more partners). The initial stage of nucleation is thus very defect sensitive.



Simple surface steps qualify as efficient "defects" for nucleation.

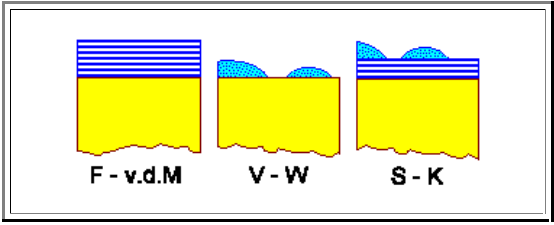
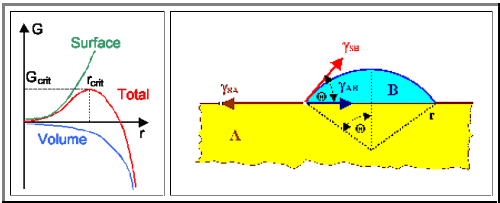
- Small deviations from perfect orientation provide large step densities. Nucleation therefore can be very sensitive to the precise {hkl} of the surface
- Intersections of (screw) dislocation lines with the surface also provide steps.
- This may cause grain boundaries and other defects in the growing layer.
- Scanning probe microscopy gives the experimental background



There is always a nucleation barrier that has to be overcome for the first "B-clusters" to form on A

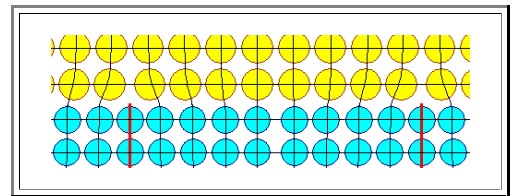
the three involved interface energies, all expressed in the "wetting angle", plus possibly some strain are the decisive inputs for the resulting growth mode.

- **Frank - van der Merve:** Smooth layer-by-layer growth
- **Vollmer - Weber:** Island growth
- **Stranski - Krastonov:** Layer plus island growth



Epitaxial layers are crucial for semiconductor technology.

- Misfit of lattice constants will produce strained layers upon epitaxial growth; strain relief happens by the formation of misfit dislocations.
- Misfit dislocations must be avoided at all costs!
- Below a usually rather small critical thickness d_{crit} of the thin layer no misfit dislocations will occur.
- Rule of thumb:
0.5 % misfit $\Rightarrow d_{\text{crit}} \approx 10 \text{ nm}$



The internal structure of thin films can be anything known from bulk materials plus some (important!) specialities.

a-Si: Micro electronics
a-Si:H: Solar cells, **LCD** displays
 $\mu\text{-Si:H}$: Solar cells

Properties of thin films can be quite different from that of the bulk material.

- The reason can be differences in length scales.
- Semiconductor technology relies to some extent on superior thin film properties.

Much better in thin films

- **Electrical break-down field strength of dielectrics.**
- **Critical current densities in conductors.**

Exercise 3.6-1

All Questions to 3