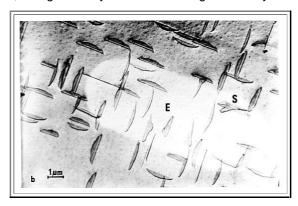
Oxidation Induced Stacking Faults in Silicon

- Oxidation of Silicon produces interstitials in supersaturation. These surplus interstitials tend to agglomerate in discs i.e. stacking fault loops. The difficult part is the nucleation; it determines what will happen. We have to consider two ways of oxidizing **Si**, we first consider
- Surface oxidation: The surface oxidizes homogeneously by exposing it to an oxidizing atmosphere at high temperatures. This is the normal oxidation process. The emission of interstitials occurs at the interface; the interstitials diffuse into the bulk; the supersaturation decreases with the distance from the surface.
 - There is no easy nucleation for an interstitial type dislocation loop as long as the interface is defect free. If defects are present, most prominent small <u>precipitates of metal impurities</u> as, e.g. **Fe**, **Ni**, **Cu**, they may serve as nucleation center for the interstitials; a stacking fault penetrating in a semicircular fashion into the bulk is formed. If many precipitates are available, a large density of small stacking faults may be observed:

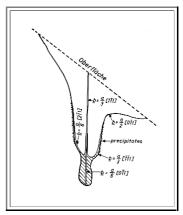


- The oval shaped area with a lighter contrast is the emitter of a bipolar transistor. In preferential etching this would look similar to what was shown as an <u>illustration for etching</u>.
- Some of these small stacking faults have a peculiar, "sailing-boat" like shape (marked by "S" in the picture above). Below, a detailed view of a "sailing boat stacking fault":

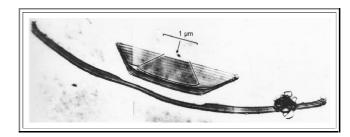


- These "sailing boats" are formed whenever the nucleation produces two stacking faults on different planes (Connected by a pair of <u>stair-rod dislocations</u>). Obviously the diffusion of interstitial down the central dislocation dipole must be rather efficient. These stacking faults penetrate through the **pn**-junction and lead to a total loss of the transistor.
- In rare cases, "sailing boat" stacking faults started to unfault. For reasons still unknown, the unfaulting process stopped at a certain depth (maybe due to doping influence?); the resulting structure is remarkable, because it contains all types of dislocations that exist in an **fcc** lattice in one defect:





- We have the perfect dislocation ($\mathbf{b} = \mathbf{a}/2 < 110 >$), the Frank partial dislocation ($\mathbf{b} = \mathbf{a}/3 < 111 >$); the Shockley partial dislocation ($\mathbf{b} = \mathbf{a}/6 < 112 >$) and the stairrod dislocation ($\mathbf{b} = \mathbf{a}/6 < 110 >$) in one defect.
- If there are only a few precipitates; they may nucleate a stacking fault many times. As soon as the first dislocation loop is too large, a new one will form. As a result, a whole system of overlapping stacking faults is seen (for every third one the contrast disappears because the sum of the displacement vectors is a lattice vector).
 - In this example the precipitate is still visible as a black dot in the center of the stacking fault system. This is usually not the case because the precipitate is incorporated into the oxide and etched off.



- If the **Si** contains some supersaturated oxygen (at high temperatures an equilibrium defect as an interstitial; "**O**i"), we may observe internal oxidation.
 - A SiO₂ precipitate forms by the agglomeration of O_i; but this may equally well be considered to be an internal oxidation of a small volume of Si. Again, interstitials are produced with the tendency for agglomeration.
 - In contrast to surface oxidation, nucleation is rather easy. The small SiO₂ precipitate, especially if it is not spherical, has a stress field that helps to nucleate the stacking fault of the interstitials. We thus find oxide precipitates surrounded by large stacking faults.
- Both processes the oxide precipitation and the stacking fault formation occur simultaneously; new precipitates may be nucleated at the Frank dislocation and vice versa.
 - In the course of several high temperature treatments; the processes start all over again and complicated structures develop:



Several perfect stacking fault loops overlap (truncated by the sample surface, one of which has been preferentially etched; the etch pits at the dislocations are clearly visible). Some of the loops serves as nucleation sites for a second and third round of oxygen precipitation (shown as small coffee-bean like contrasts).